

HETEROSTRUCTURE P-N JUNCTIONS (optelhd youtube video)

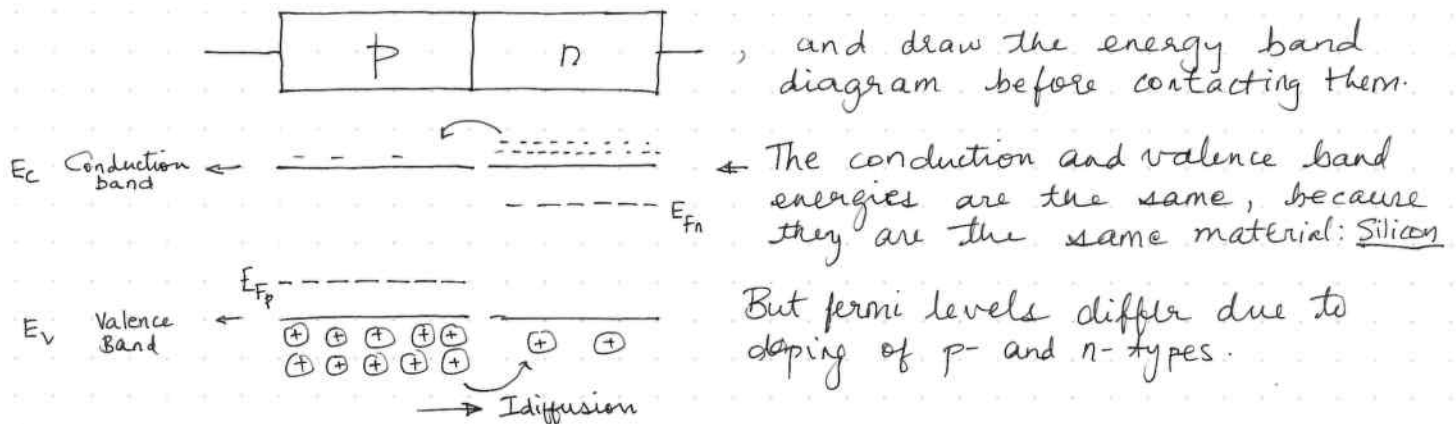
- Prof M.R. Sheroy, IIT Delhi, Physics

PN junctions are of types:

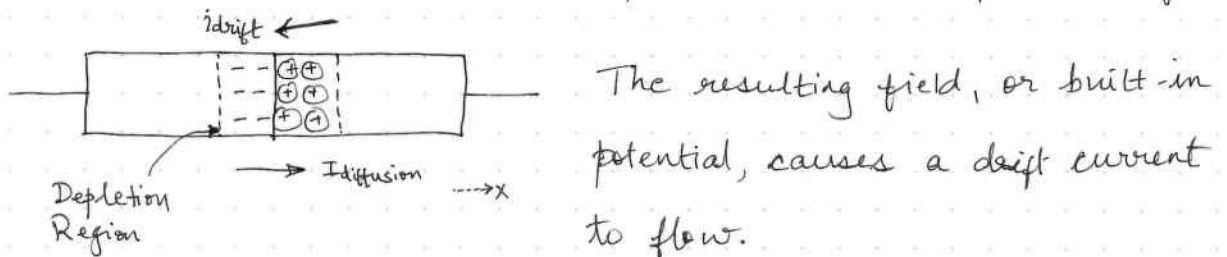
- 1) HOMOJUNCTION: A junction between same material, but different doping concentrations / types. Eg: p-type / n-type silicon.
- 2) HETEROJUNCTION: A junction between two different materials with different bandgaps.
- 3) SCHOTTKY JUNCTION: A junction between metal and semiconductor.

Let us go through a brief review of homojunction band energy diagrams as it will lay the foundation for heterojunctions.

Consider a homojunction between a p- and n-type silicon material:



There are a large number of electrons on n-side, and a large number of holes on p-side that will diffuse over to the other side due to concentration gradient, leaving behind holes in the n-side and electrons in the p-side, creating depletion region.



At equilibrium: $i_{diffusion} + i_{drift} = 0$, and it can be shown

that $\frac{dE_f}{dx} = 0 \Rightarrow E_f$ is independent of x . E_f is a constant.

Fermi level is a constant

BAND BENDING

Now, let us consider the energy levels when pn junction is in equilibrium.

Let $\rho(x)$ be the 1-D charge density in the depletion region.

According to Gauss' law:

$$\nabla \cdot E = \frac{\rho(x)}{\epsilon}$$

In 1-D: $\frac{dE}{dx} = \frac{\rho(x)}{\epsilon}$

or $E(x) = \int \frac{\rho(x) dx}{\epsilon}$

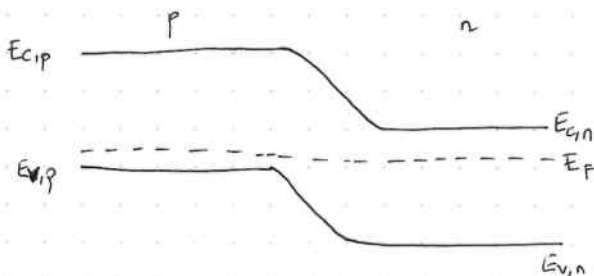
Calculating potential: $E = -\frac{dV}{dx}$

or $V(x) = -\int E(x) dx$

And Energy $E(x) = (-e)V(x)$

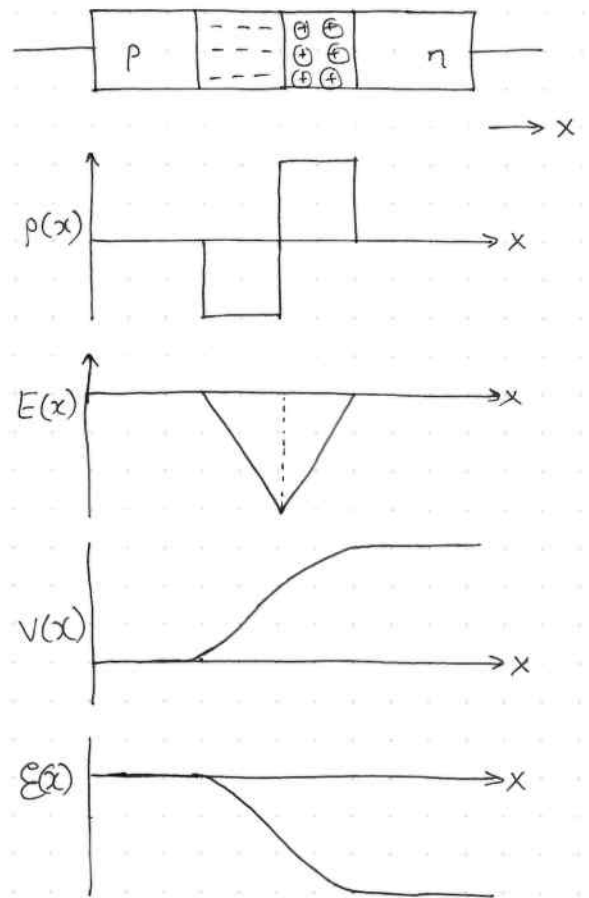
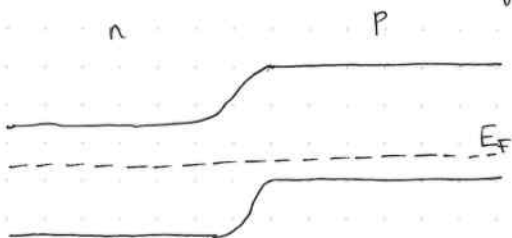
The migration of charges across the junction has caused the energy to vary across the depletion region.

Thus, the bands of energy are said to "bend" to accommodate this continuous variation of energy. Now, we are able to draw the band diagram of a p-n homojunction.



← This is obtained by adding $E(x)$ to the energy diagrams of the p & n type materials.

Conversely, the same line of reason is applicable to an n-p junction

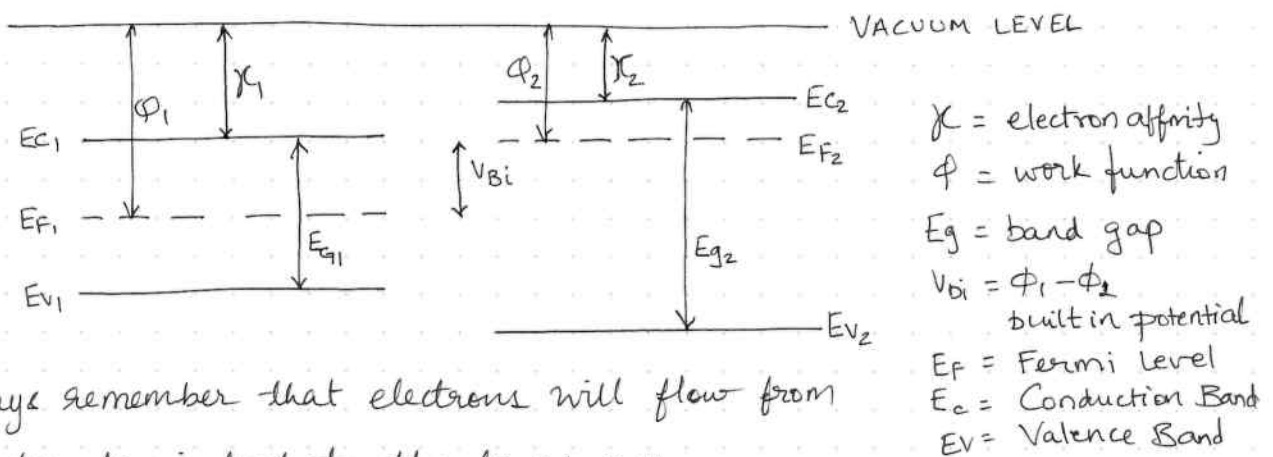


ANDERSON'S RULE:

Before we go deep into constructing energy diagrams for heterojunctions; we should understand Anderson's rule.

When constructing an energy band diagram, the vacuum levels of two semiconductors on either side of the heterojunction should be aligned.

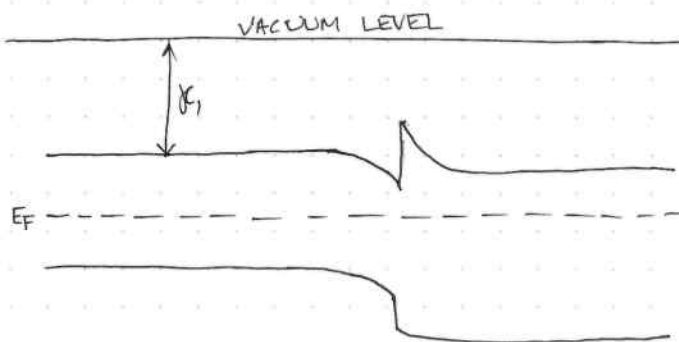
Once vacuum levels are aligned, electron affinity (energy required to free an electron from conduction band) and band gap values can be used to construct an energy diagram.



Always remember that electrons will flow from the higher fermi level to the lower one.

Knowing this, you can determine the charge distribution at equilibrium and use that to calculate E , V , and \mathcal{E} , like the case of a homojunction p-n, earlier.

In the above example, electrons will flow from 2 to 1, resulting in the following:



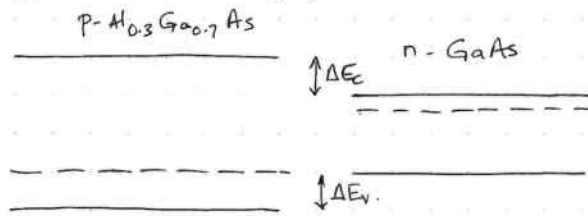
χ and E_g IN

Common Semiconductors	χ (eV)	E_g (eV)
Si	4.05	1.12
Ge	4.0	0.66
GaAs	4.07	1.43
InP	4.35	1.35
AlAs	2.62	2.16

HETEROJUNCTIONS

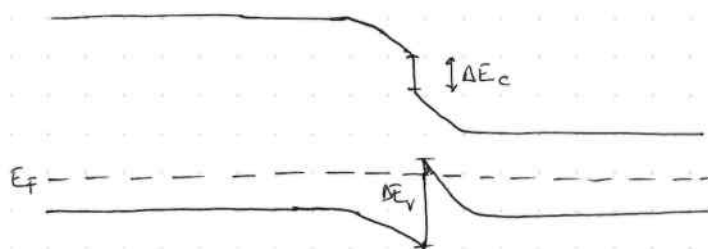
(nptelhad youtube video continued)

In a heterojunction, we contact two dissimilar materials with different bandgaps. Consider an example of p-AlGaAs & n-GaAs.



Electrons will flow from n-GaAs to p-AlGaAs, making the depletion region negative in AlGaAs and positive in GaAs.

As a result, the n-GaAs bands will bend down till Fermi level aligns. However, due to the bandgap differences in E_c and E_v , there will be an abrupt energy change at the interface as shown below:

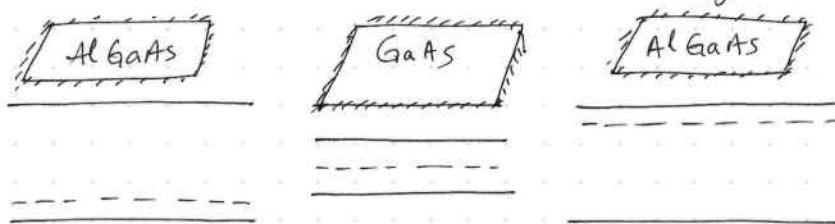


This shows energy diagram at equilibrium.

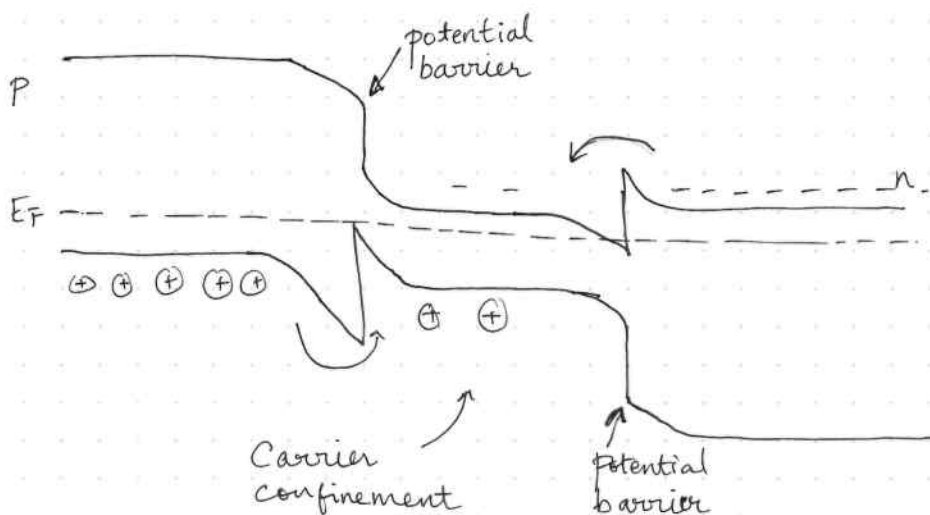
Band diagrams for other heterostructures can be drawn in a similar way.

For a brief moment, consider double heterostructures, because there is something fascinating that happens in terms of energy.

Consider a AlGaAs - GaAs - AlGaAs system:



Thickness of sandwiched GaAs layer is in our control



If this junction is forward biased, n-side energy moves up by $E = (-e)(-V)$, spilling electrons and holes in the GaAs layer. But these carriers are confined by the presence of potential barrier \Rightarrow high charge density